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M.Sc. RNLKWC-/CEM-103/22

2022

**INORGANIC CHEMISTRY**

**M.Sc. First Semester End Examination - 2022**

**PAPER - CEM-103**

**Full Marks : 40**

**Time : 2 hours**

*The figures in the right-hand margin indicate marks.  
Candidates are required to give their answers in their own  
words as far as practicable.  
Illustrate the answers wherever necessary.*

**Group-A**

1. Answer any four questions 2×4=8
- (i) Show that  $C_{2h}$  is an Abelian group.
  - (ii) What do you mean by cooperative interaction in  $O_2$  affinity of Hemoglobin?
  - (iii) Prove that :  $(ABCD)^{-1} = D^{-1}C^{-1}B^{-1}A^{-1}$

*(Turn Over)*

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- (iv) What is ceruloplasmin?  
(v) Find out the inverse of  $S_6^2$ - operation.  
(vi) What is Wilson's disease?

**Group - B**

2. Answer any four from the following . **8×4=32**

- (i) a) Discuss the structural features of Hemerythrin (Hr) and hemocyanin (Hc). Write their role in oxygen transport process.  
b) Write the names of two Zn containing enzymes. Discuss their functions. 4+4
- (ii) a) Write short note on alzheimer disease.  
b) Briefly explain the biofunction of ferredoxins. 4+4
- (iii) a) Find the point group of the following molecules/ ions.  
 $SO_3^{2-}$ ,  $cis[Pt(NH_3)_2Cl_2]$ ,  $trans[Cr(H_2O)_4Cl_2]$   
b) Show that if there are two  $C_2$  axes in a group, there must be a third  $C_2$  axis which is perpendicular to other  $C_2$  axes.

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- c) Write all symmetry operations present in  $B_2H_6$  molecule. Is it a abelian or non-abelian group? Explain. 3+2+3
- (iv) a) Write short note on "Great Orthogolality Theorem".  
b) If 'A' is conjugated with 'B' and 'A' is conjugated with 'C', then prove that 'B' & 'C' are mutually conjugate to each other.  
c) Derive the matrix form of  $S_n(Z)$  operation. Write the charater of  $S_3(Z)$  operation. 3+2+3
- (v) a) A beam of X-rays of wavelength 0.071 nm is diffracted by (110) plane of rock salt with lattice constant of 0.28 nm. Find the glancing angle for the second-order diffraction.  
b) Write with explanation whether the reflections for the plans (111), (200) and (1.35) are allowed for SCC, BCC and FCC lattices.  
c) Discuss Edge dislocation with diagram. 2+3+3

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- (vi) a) Write the difference between Schottky and Frenkel defect.
- b) How does the temperature influences the electrical conductivity of metal and semiconductor?
- c) Write the differences between type I and type II semiconductor. 2+3+3
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