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B.Sc. RNLKWC(A)-/DSE-4T/22

2022

CHEMISTRY

[Honours]

B.Sc. Sixth Semester End Examination - 2022
PAPER - DSE-4T

Full Marks: 40

Time: 2 hours

The figures in the right-hand margin indicate marks.

Candidates are required to give their answers in their own words as far as practicable.

Illustrate the answers wherever necessary.

Group - A

1. Answer any five from the following: $5\times 2=10$

- (a) What types of molecular forces are present in polymers?
- (b) What are the physical significance of the value p=0 and p=1 where p=extent of reaction.
- (c) Defind chain growth polymerization process.

(Turn Over)

(d)	10g of organic substance when dissolved in two litres of			
	water gave an osmotic pressure of 0.59 atm at 7°C.			
	Calculate the molecular weight of the substance.			

- (e) What are the differences between free radial an ionic polymerization?
- (f) The glass transition temperature of poly stryrene is higher than that of polypropylene and polyethylene. Explain why?
- (g) What is back bitting?
- (h) Write down the criteria of a polymer that required for crystallisation.

Group - B

Answr any four from the following.

 $4 \times 5 = 20$

- 2. (a) Derive Carothuss equation for bifunctional monomer. 3
 - (b) Explain the structure dependent conductivity of polymer? 2
- 3. (a) What is free volume and how it originates into polymer?
 - (b) Derive expression of enthalpy change (ΔH) for polymer mixing.2

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(Continued)

4.	Cla	assity polymer with examples. 5		
5.	(a)	What is the number average molecular weight and weight average molecular weight? How they are related to each other?		
	(b)	Draw a Dilatometric plot (specific volume vs temperature)		
6.	(a)	How fluro polymers are prepared?		
	(b)	Discuss photo chemical polymerization process.		
7.	(a)	What is polydispessity index? Cite one example where its value is unity.		
	(b)	What is the difference between polymer and macro		
		molecule?		

Group - C

Answer any one from the following:

 $1 \times 10 = 10$

8. Write short nots on : (any four)

 $2\frac{1}{2} \times 4 = 10$

- (a) Zieigller-Natta catalyst.
- (b) Electrochemical polymerization process.

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(Turn Over)

- (c) Flory-Hiiggins equation.
- (d) Polyurethane Polymer.
- (e) Polymer additives.
- (f) Co-ordination polymerization.
- (a) A polymer sample consists of a mixture of three mono-dispense polymers with molar mares 250000, 300000 and 350000 g/mol in the ratio 1:2:1 by number of chains. Calculate Mn; Mw and Mw/Mn.
 - (b) Write the differences between bulk polymerization and solution polymerization.
 - (c) What are the advantages and disadvantages of emulsion polymerization.